

# Application of Computer in Chemistry

## SSC 3533

### MULTIVARIATE CALIBRATION

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# Multiple Linear Regression

$Y$  is dependent on more than one  $x$ 's

## Model

$$y = \beta_0 + \beta_1 x_1 + \beta_2 x_2 + \dots + \beta_m x_m$$

$$y = b_0 + b_1 x_1 + b_2 x_2 + \dots + b_m x_m$$

# Example

Absorbance of a mixture of more than one components is determined by the concentration of each component in the sample:

$$A = b_1C_1 + b_2C_2 + b_3C_3$$

A – absorbance

b – molar absorbtivity

c – concentration

# Calculation Method

- The same method used to get simple linear equation is also used here i.e. least squares method
- Find an equation that minimize the sum squared deviation.

$$Q = \sum_{i=1}^n (y_i - \hat{y}_i)^2$$

- To simplify the calculation, matrix operation is used to obtain  $b_0, b_1, b_2 \dots b_m$

# Matrix Solution

The Y and X variables are expressed in matrix form

$$Y = \begin{bmatrix} y_1 \\ y_2 \\ \cdot \\ \cdot \\ y_n \end{bmatrix} \quad X = \begin{bmatrix} 1 & x_{11} & x_{12} & \cdot & \cdot & x_{1,p-1} \\ 1 & x & x_{22} & \cdot & \cdot & x_{2,p-1} \\ \cdot & \cdot & \cdot & \cdot & \cdot & \cdot \\ \cdot & \cdot & \cdot & \cdot & \cdot & \cdot \\ 1 & x_{n,1} & x_{n,2} & \cdot & \cdot & x_{n,p-1} \end{bmatrix} \quad b = \begin{bmatrix} b_0 \\ b_1 \\ \cdot \\ \cdot \\ b_{p-1} \end{bmatrix}$$

# The regression equation

$$\hat{Y} = Xb$$

$$b = (X'X)^{-1}X'Y$$

# Determination of a mixture

Example:

The absorbance of a mixture of  $\text{Cl}_2$  and  $\text{Br}_2$  in chloroform was measured at various wavelengths using a UV-Visible Spectrophotometer.

Based on the following data, determine the concentration of chlorine and bromine in the mixture.

# Data

Wave Number	Molar absorbtivity (k)		Absorbance (A)
	Cl <sub>2</sub>	Br <sub>2</sub>	
22	4.5	168	34.10
24	8.4	211	42.95
26	20	158	33.55
28	56	30.0	11.70
30	100	4.70	11.00
32	71	5.30	7.98



# Multivariate Calibration

- In Chemistry, we often encounter determination of more than one parameters at a time.
- Not only in  $X$  but sometimes in  $Y$  too!!

# Determination of protein and water in wheat using near-IR

- Objective: To determine protein and water content in wheat using near-IR.

## Calibration

- Get IR spectra for 3 samples of wheat
- Measure 'near-IR reflectance' at 8 wavelengths
- **Data matrix:  $X (3 \times 8)$**
- Determine, using normal method, percentage of protein (Kjeldahl method for N) and percentage of water in the three samples of wheat.
- **Concentration matrix:  $C (3 \times 2)$**

# Data Matrix \*

$$\mathbf{X}_{3 \times 8} = \begin{bmatrix} 2 & 1 & 2 & 1 & 1 & 5 & 3 & 1 \\ 1 & 2 & 2 & 3 & 7 & 4 & 7 & 1 \\ 1 & 2 & 1 & 3 & 6 & 2 & 6 & 2 \end{bmatrix}$$

$$\mathbf{C}_{3 \times 2} = \begin{bmatrix} 2 & 4 \\ 3 & 5 \\ 8 & 9 \end{bmatrix}$$

\* Note, these are not real values

# Multiple Linear Equation

$$\hat{C} = X.B$$

$$B = (X'.X)^{-1}.X'.C$$

Notice that C is now the Y variable and Absorbance is now X the variable – **Inverse Calibration** – why?

# Prediction

$$\mathbf{C}_{\text{unk}} = \mathbf{X}_{\text{unk}} \mathbf{B}$$

- Make measurement on unknown samples
- Use the equation and stored value of **B** to Calculate **C** for the unknown samples

# Summary

## Calibration

- Using standards, measure concentration – **C**
- Obtain spectrum of standards – **X**
- Develop model: **C = XB** - store **B**

## Prediction

- Obtain spectrum of unknown – **X<sub>unk</sub>**
- **C<sub>unk</sub> = X<sub>unk</sub> B**

# Problems with MLR

- Only suitable in ideal conditions:
- Strong correlation between X variables can cause problems.
- Number of samples must be greater than number of variables (Five to one rule)



**Need a more robust method!!**

# Principal Component Regression

- Original variable,  $X$ , transformed into a new set of variables called principal components, PC
- Principal components, by definition, are not correlated with each other (orthogonal)
- Number of variables can be reduced – take only the important factors



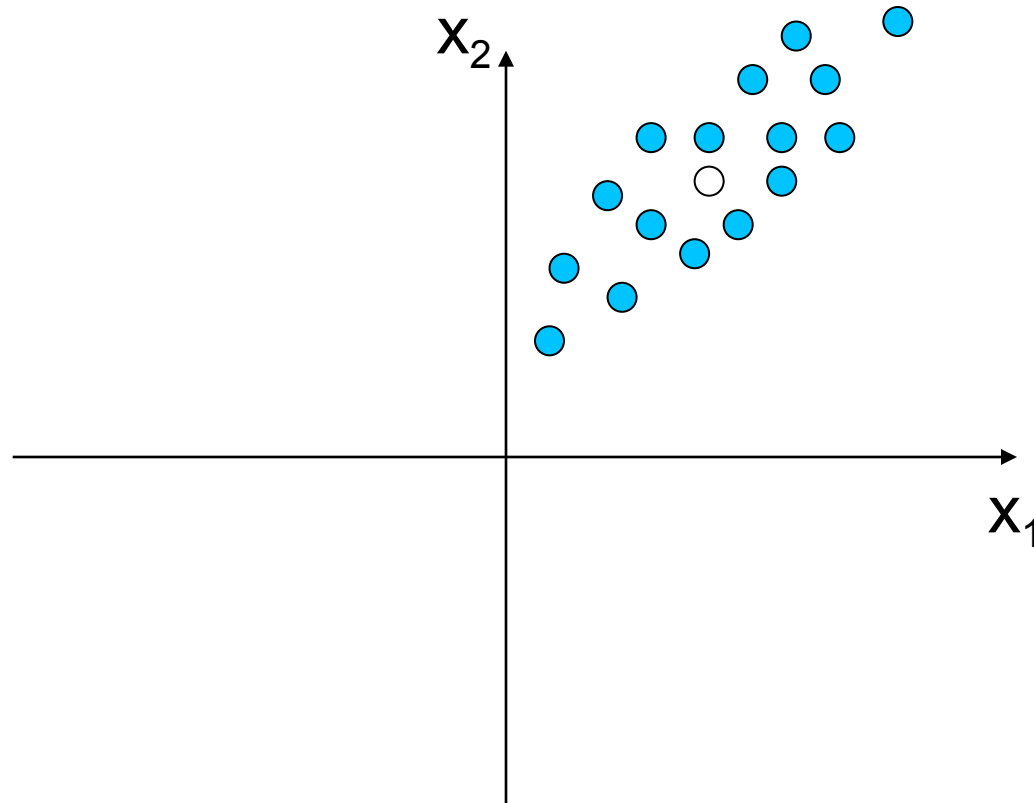
# Principal Components Analysis

- Theory and original idea in mathematics and physics (1800s) – eigen analysis
- First applied in psychometrics (1930s, 1940s)
- Also called factor analysis
- Became popular in Chemistry 1970s

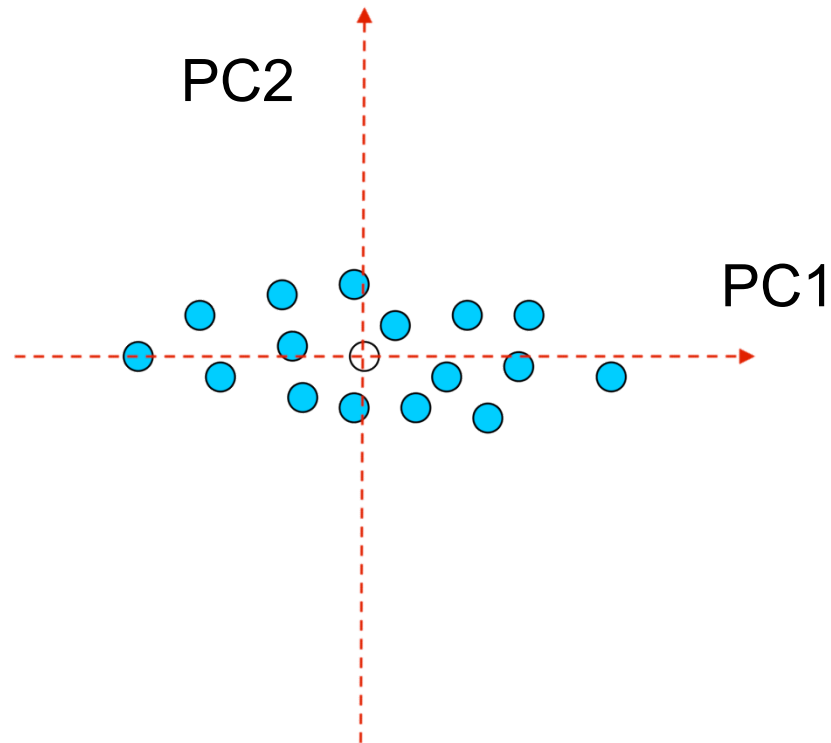
# Concept of PCA

- Transformation from original coordinates into new coordinates
- The new coordinates are selected in such a way so that the highest variance is along PC1
- PC2 is at right angle to PC1

# Original Data



# Transformed data



# PCA

$$X = T \cdot P + E$$

X – Original data (I x J)

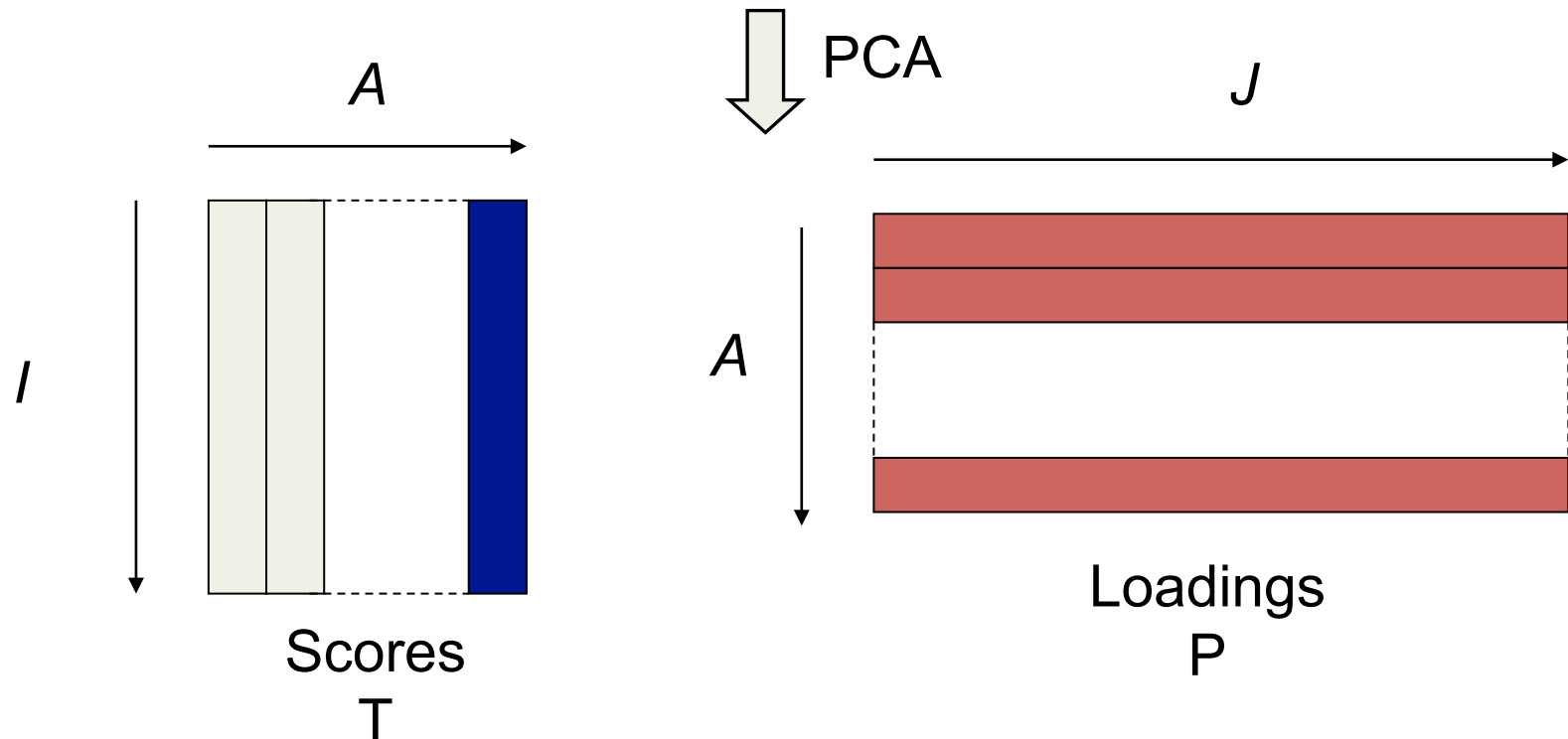
T – Scores ( I x A)

P – Loading (A x J)

E – Error

# Example

- Chromatogram obtained from HPLC using diode array detector
- Experimental Data:
  - Sampling 30 points (every 1 s)
  - At 28 wavelengths (interval 5 nm)
  - Absorbance Data (AU)
- Data matrix,  $X$  (30 x 28)



# Summary of PCR

1. Get principal components using PCA

$$X = T \cdot P + E$$

2. Regress C on T

$$C = T \cdot R + E$$

3. Prediction:

$$C_{unk} = T_{unk} \cdot R$$



# Example of PCR

- Mixture of 10 polycyclic aromatic hydrocarbon compounds (PAH)
  - Pyrene, Acenaphthene, Anthracene, Acenaphthylene.....
- 25 EAS Spectra at 27 wavelengths (from 220 – 350 nm at 5 nm intervals)

# Partial Least Squares Regression, PLS

PCA is performed not only on X block but also on C block

$$X = T \cdot P + E$$

$$C = U \cdot Q + F$$

T – Score for X

P – Loading for X

U – Score for C

Q – Loading for C

# PLS

- In ideal conditions, factors in X and factors in C should be the same.
- In reality, they are different. Get the relationship between U and T

$$u = bt + e$$

